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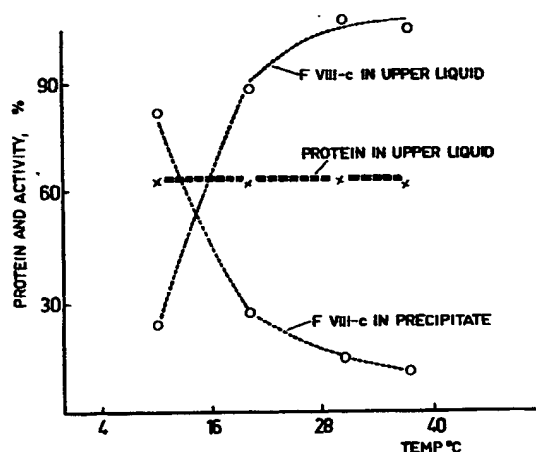
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54 A process in purification and concentration of the factor VIII-complex.

57 A process in purification and/or concentration of factor VIII complex, starting from a preparation of factor VIII complex, obtained and concentrated in a known manner, mostly in the form of a precipitate, such as cryoprecipitate or Cohn's fraction I-0. This preparation is dissolved in a glycine solution of at least 1.5 M at a temperature of at least +15° C and pH of 6.3-7.8 and a supernatant liquid is recovered as a product or for further working-up.



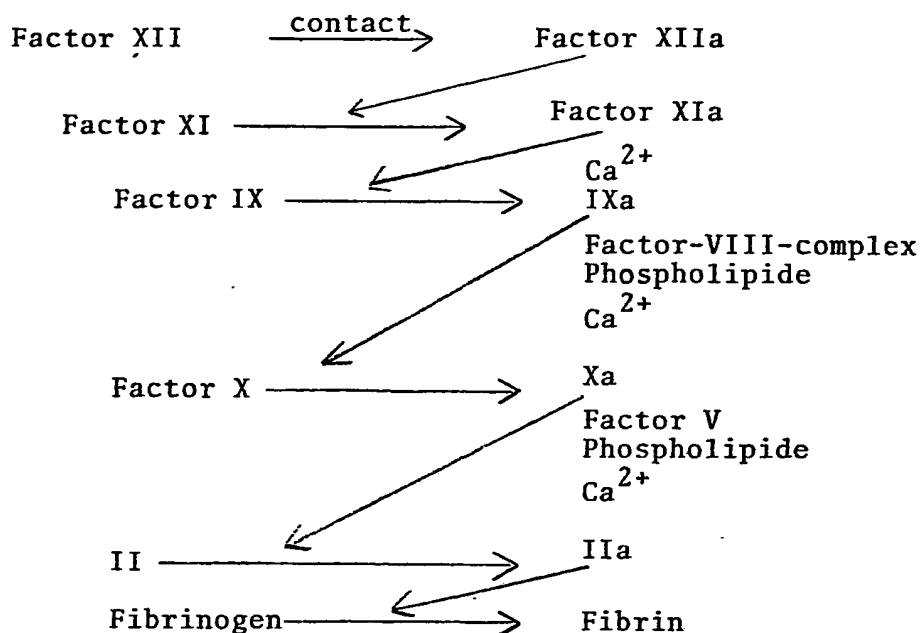
**EP 0 032 655 A1**

A process in purification and concentration of the factor  
VIII-complex

This invention relates to a process in purification and/or concentration of the factor VIII complex. It is started from a preparation prepared in a manner known per se, in which the factor VIII complex is enriched. The  
5 preparation is mostly in the form of a precipitate.

The factor VIII complex is lacking or its activity is reduced in hemophilia of type A and in v. Willebrand's disease. The symptoms in these diseases are often serious bleedings in joints and muscles and from mucous membranes.

10 The factor VIII complex takes part in the biochemical reactions promoting the coagulation of blood. In the coagulation of blood, an insoluble polymer, fibrin, is formed from a soluble protein, fibrinogen. The reason for the polymerization or formation of fibrin is an enzymatic  
15 change of the fibrinogen molecule, which is caused by the enzyme trombin (factor IIa). This enzyme is formed from protrombin (factor II) under the influence of an enzyme, called factor Xa. Said enzyme is also present as a zymogen in the blood before the coagulation. The zymogen form is  
20 called factor X. The conversion of factor X to factor Xa also takes place enzymatically by means of an enzyme called factor IXa. The factor VIII complex takes part in this reaction (as a co-factor) together with calcium and phospholipide. In the following scheme the most important reac-  
25 tions in the coagulation (clotting) of blood are illustrated. Totally a great number of various factors take part here.



The factor VIII complex consists of at least two components, one of which is called the factor VIII:C, in which C means that this component is responsible for the coagulation activity of the complex in the reaction chain shown. This component is considered to contain the antigen proved by means of antibodies, which are developed in certain persons suffering from hemophilia and which prevent the coagulation activity of factor VIII:C. The antigen is called F VIII:CAG. The other component has been called factor VIII-RAG or F VIII-related antigen. This antigen is different from the antigen F VIII:CAG. The factor VIII:C and the (antigen) factor VIII:CAG are lacking in hemophilia of type A in a serious form. In this disease there is a normal content of factor VIII-RAG. In v. Willebrand's disease, there is a lack of factor VIII-RAG in the blood and a corresponding lack of factor VIII:C. For persons suffering from a serious form of v. Willebrand's disease the lack of factor VIII-RAG is almost total and the content of factor VIII:C is about 5 % of the normal content. In v. Willebrand's disease, the activity of the so-called factor VIII:RCF is highly reduced. This activity is an expression of a component in

plasma which causes agglutination of trombocytes in the presence of the antibiotic "Ristocetin". It is now considered that the activity of the factor VIII:RCF is an expression of the component or components in plasma which has (have) the factor VIII-related antigen (factor VIII: RAG). The lack of factor VIII:RCF in v. Willebrand's disease has been found to be correlated to the prolonged capillary bleeding time existing in this disease and which is an expression of a defective trombocyte function. This function is normal in hemophilia type A, whereas the coagulation time of the blood is prolonged due to the reduced content of factor VIII:C. In v. Willebrand's disease, the coagulation time is also prolonged as there is also a lack of factor VIII:C in this disease, especially in serious forms thereof. The lack of factor VIII:C in v. Willebrand's disease is considered to be a consequence of the lack of factor VIII:RAG/RCF, which seems to act as carrier molecule for factor VIII:C. In accordance with the existing values, one can illustrate the factor VIII-complex schematically in the following way:

$$F \text{ VIII:RAG/RCF} = = F \text{ VIII:C/CAG}$$

A number of processes are known at present for the preparation of plasma concentrates for clinical use, which contain the factor VIII complex or parts thereof. The complex can be precipitated from Cohn's fraction I with alcohol. Additional concentration of the complex can be carried out by extraction of inert protein with about 1 M glycine solution in cold. Fractionation of plasma with ether or tannic acid has also been used for the preparation of factor VIII concentrates. Concentrates prepared by precipitation in cold of the factor VIII complex have been widely used due to the simplicity of the method. In certain cases, polyethylene glycol has been used for precipitation of the factor VIII complex. Variants of the glycine method, in which either other amino acids than glycine have been used or glycine has been used for precipitation of the factor VIII complex are also known.

Concentrates prepared according to the glycine method contain all factors in the factor VIII complex, but the specific activity is low, and therefore great volumes of solution must be injected in the treatment of hemophilia.

5 Preparations with a high specific activity have often been found to lack the factor correcting the prolonged bleeding time in v. Willebrand's disease.

As to the yield of the coagulation active part (F VII:C), this is low in most preparation processes or

10 between 20 and 30 % at best.

In the treatment of hemophilia, it is of a great importance to use concentrates of the F VIII complex with a high specific activity of all the factors included in the complex. Hemophilia A as well as v. Willebrand's disease

15 can be treated with such a concentrate. A specific activity which is 200-300 times greater than that in plasma also permits preparations containing the necessary therapeutic dose in a small volume (5-10 ml). This makes the treatment easier as the dose can be administered in an injection

20 syringe. This also makes it possible to carry out the treatment for example at home by the sick person himself as is usual in the treatment of diabetes with insulin.

In most countries, the supply of blood is unsatisfactory in respect of the possibilities of preparing sufficient

25 amounts of factor VIII complex for treatment of hemophilia from available blood volumes. This situation might be considerably improved if the percentage yield of the factor VIII complex could be increased in preparation from blood.

It has now been found to be possible to prepare a

30 highly satisfactory factor VIII complex having a specific activity that is about 150-250 times greater than that in plasma, while the yield is between 40 and 60 %. The process according to the invention is characterized in that the resulting preparation of factor VIII complex is dissolved in

35 a glycine solution of at least 1.5 M at a temperature of at least +15°C and a pH of 6.3-7.8, preferably pH 6.8, and that supernatant liquid is recovered as a product or for

further working-up. The resulting product (factor VIII complex) contains the antihemophilic factor (factor VIII:C/CAGL) and the component (factor VIII-RAG/RCF) which is lacking in v. Willebrand's disease.

5       The starting preparation obtained in a way known per se can be prepared by cryoprecipitation or fractionation of plasma with alcohol. Through the process of the invention, impurities are then precipitated in a glycine solution at and about room temperature and at a neutral  
10 pH. In previously known processes, glycine solutions have been used at a low temperature for dissolution of impurities from a precipitated factor VIII complex. Glycine at a concentration of about 2 M has also been used for precipitation of the factor VIII complex from  
15 a protein solution at a low temperature.

      According to the invention, the preparation of factor VIII complex is now treated at a temperature of at least 15°C, preferably at a temperature of 15-37°C, such as 20-35°C, and most preferably at about 30°C. The  
20 temperature should not normally exceed 40-45°C.

      In doing this, the factor VIII complex remains in solution, while contaminating protein is precipitated. The pH is approximately neutral. The ionic strength of the solution should be between 0.1 and 0.5, calculated  
25 as buffer salts. The glycine concentration at the precipitation of impurities can vary between 1.5 M and a saturated solution and is preferably 1.6-2.3 M, such as 1.8-2.2 M, and most preferably about 2.0 M. By this precipitation of protein which is inert in this connect-  
30 ion, the specific activity in the supernatant liquid is increased by 2-7 times and the yield is between 80 and 100 %.

      The invention is described more in detail in the following examples with reference to its use in the  
35 fractionation of blood plasma. In the examples, it is referred to the enclosed drawing, in which Fig. 1 shows protein and activity in the supernatant liquid and the specific activity as a function of the molarity of the

glycine solution. Fig. 2 shows the activity in supernatant liquid and in precipitate and protein in the supernatant liquid as a function of the temperature. In Figs. 3 and 4 the same quantities are shown as a function of pH and ionic strength, respectively.

Example 1. Blood is collected in citrate-phosphate-dextrose-adenine solution in an amount of 50 ml of citrate solution per 450 ml of blood. Blood cells are removed by centrifuging and plasma is sucked off. The plasma is fractionated with alcohol according to Cohn's fractionation method. Fraction I is washed in cold with glycine solution. The remaining precipitate, called fraction I-0, contains factor VIII complex and is dissolved in sodium citrate solution (pH 6.8) of 0.055 M to a protein concentration of about 2 %. 8 parts of a buffer with a pH of 6.8 and containing varying amounts of glycine, 0.125 M NaCl and 0.025 M imidazole are added to 4 parts of this solution at room temperature. After stirring for 15 minutes the precipitate formed is removed by centrifugation at ambient temperature. The precipitate is dissolved in citrate solution. The precipitate as well as the supernatant liquid is analyzed in respect of activity of factor VIII:C (normalization of the coagulation time of plasma from persons with hemophilia A after recalcification) or with a commercial reagent (IMCO AB, Stockholm). The total protein is determined by means of Lowry's method. The results of this test appear from Fig. 1. As shown in the figure, considerable amounts of inert protein are precipitated from the solution of fraction I-0 at glycine concentrations of 1.1-2.2 M. When the glycine content increases over 2.2, additional precipitation of inert protein takes place, but considerable losses of factor VIII:C can occur from the supernatant liquid. At a glycine content of 1.5-2.0 M, the yield of factor VIII:C in the supernatant liquid is high, while the content of protein therein is much lower than in the original solution. The specific activity of the supernatant liquid is about

5 times higher than in the original solution at a glycine concentration of 2 M.

Example 2. Blood and plasma are prepared in the way described in example 1. A counterpart of fraction I-0 is prepared by freezing of the factor VIII complex by adding polyethylene glycol 4000 in an amount of 1 % to the plasma. The mixture is frozen at  $-70^{\circ}\text{C}$ . At a slow thawing to  $+4^{\circ}\text{C}$  a precipitate called cryoprecipitate remains when all ice has melted. This precipitate contains between 60 and 90 % of the factor VIII complex of the plasma. The precipitate is dissolved in 0.055 M sodium citrate solution (pH 6.8) to a protein content of about 3 %. 10 parts of a buffer containing 3.0 M glycine, 0.125 M NaCl and 0.025 M imidazole (pH 6.8) are added to 5 parts of this solution at  $10^{\circ}$ ,  $20^{\circ}$ ,  $30^{\circ}$  and  $37^{\circ}\text{C}$ . The mixtures are brought to equilibrium at the different temperatures under stirring for 15 min. The precipitate formed is removed by centrifugation. Precipitate and supernatant liquid are analyzed as in example 1.

As is apparent from Fig. 2, the separation of the factor VIII complex from inert protein is highly dependent on the temperature of the glycine-protein mixture. At temperatures above  $15^{\circ}\text{C}$ , the main part of the factor VIII complex is found in the supernatant liquid, while at lower temperatures, the main part is in the precipitate.

Example 3. The cryoprecipitate (see example 2) is dissolved in a 0.055 M citrate solution (pH 6.8) to a protein content of about 3 %. 8 parts of buffer containing 3 M glycine, 0.125 M NaCl and 0.025 M imidazole, the pH of which has been adjusted to 6.3; 6.8; 7.3 and 7.8, respectively, are added dropwise to 4 parts of this solution. After stirring for 15 min at  $30^{\circ}\text{C}$ , the precipitate formed is removed by centrifugation and dissolved in citrate solution (pH 6.8). The supernatant liquid and the precipitate are analyzed according to example 1.

As is evident from Fig. 3, the differences in activity and yield at different pH are relatively small. An optimal



separation of the factor VIII complex on behalf of supernatant liquid is obtained at approximately neutral pH.

Example 4. 10 parts of glycine-imidazole solution with a varying content of common salt were added dropwise to 5 parts of the cryoprecipitate (see examples 2 and 3). The composition of the different additional solutions is as follows:

1) 3 M glycine, 0.025 M imidazole, pH 6,8; 2) 3 M glycine, 0.025 M imidazole, 0.3 M NaCl, pH 6.8; 3) 3 M glycine, 0.025 M imidazole, 0.6 M NaCl, pH 6.8.

After achieving equilibrium at 30°C for 15 min the resulting precipitates are removed by centrifuging. The supernatant liquid and the precipitate are analyzed according to example 1.

The results are shown in Fig. 4. The variation in ionic strength between 0.1 and 0.5 has only a small effect on the distribution of the factor VIII complex in supernatant liquid and precipitate.

Example 5. Fraction I-0 is dissolved in citrate solution (see example 1) to a protein content of about 2 %. 10 parts of 3 M glycine containing 0.125 M NaCl and 0.025 M imidazole, pH 6.8, are added to 5 parts of the resulting solution. After stirring at 20°C the precipitate (F I) is removed by centrifugation. The supernatant liquid (upper liquid I) is cooled to 4°C. Aqueous polyethylene glycol 6000 of 30 % is then added to a final concentration of 10 %. After stirring for 30 min at 4°C, the precipitate (F II) is removed from supernatant liquid (upper liquid II) by centrifuging. All supernatant liquids and precipitates are analyzed in the way described in example 1.

The distribution of the protein and the factor VIII complex in the different fractions is apparent from the following table 1. As is apparent from the table, the fraction I-0 is suitable as a starting material for fractionation. The results show, which is most important, that the factor VIII complex in the super-

natent liquid can be concentrated by precipitation with polyethylene glycol after precipitation of inert protein with glycine.

Table 1

Fraction	Vol. mls	Protein mg/ml	Total protein mg	VIII:C Σ/ml	Total Σ.	VIII:C %	Specific activity Σ/mg
F I-0	18	19.8	357	100	42.5	100	0.12
Upper liquid I	50	-	-	0.66	33.0	77.6	-
F I	19	11.5	219	61.3	9.3	21.6	0.04
Upper liquid II	73	-	-	0.14	10.2	24.0	-
FII	3.7	16.6	61	17.1	22.0	51.8	0.36

Example 5. 10 parts of solution containing 3 M glycine, 0.125 M NaCl and 0.025 M imidazole at pH 6.8 are added to 5 parts of solution of the cryoprecipitate (see example 4) at 20°C. After stirring for 10 min the precipitate (F I) formed is removed by centrifuging. The supernatant liquid ( Upper liquid I) is cooled to 4°C, after which an aqueous solution of 30 % of polyethylene glycol 6000 is added dropwise to a concentration of 6 %. After stirring for 30 min the precipitate (F II) is removed from the supernatant liquid (upper liquid II) by centrifugation. All fractions are analyzed in the way described.

The results of the test appear from table 2. It is apparent from this that cryoprecipitate as well as fraction I-0 is suitable for this type of fractionation. The test also shows that the factor VIII complex, which according to the invention has been obtained in the supernatant liquid, can be concentrated by precipitation with polyethylene glycol.

Table 2

Fraction	Vol. mls	Protein mg/ml	Total protein mg	VIII:C g/ml	Total g	VIII:C %	Specific activity g/mg
Cryo	18.5	26.4	488	5.57	103.0	100	0.21
Upper liquid I	53	-	-	1.72	91.2	88.5	-
F I	22.5	9.6	216	0.63	14.2	13.8	0.07
Upper liquid II	31	-	-	0.09	2x 2.8	5.4	-
F II (H)	8.8	3.2	2x 28.2	3.77	2x 3.2	64.5	1.18

Example 7. Cryoprecipitate (see example 2) is dissolved in 0.055 M citrate solution, pH 6.8, to a concentration of 3 %. 3.3 parts of a solution containing 2.6 M glycine and 0.025 M imidazole, pH 7.3, are added to 1 part of this solution. The temperature of the mixture is simultaneously brought to 7°C. After the addition, the mixture is brought to equilibrium under stirring at 7°C for 15 min. A precipitate (F I) is formed and removed by centrifugation. The supernatant liquid (Upper liquid I) is saved for analysis. The precipitate is dissolved in 0.055 M citrate solution, pH 6.8, at 30°C to about the same volume as the original cryoprecipitate, after which 3.3 volumes of a solution containing 2.6 M glycine, 0.3 M NaCl and 0.025 M imidazole, pH 6.8, are added. Stirring is carried out at 30°C for 20 min. The precipitate (F II) formed is removed by centrifugation. The supernatant liquid (Upper liquid II) is adjusted to pH 7.5 with a weak sodium hydroxide solution after cooling to 4°C. An aqueous solution of 30 % polyethylene glycol 6000 is then added dropwise to a concentration of 6.5 %. The mixture is stirred at 4°C for 30 min, after which the precipitate formed is removed by centrifuging. The precipitate (F III) is dissolved in 0.055 M sodium citrate solution containing 0.055 M imidazole, pH 7.40. The supernatant liquid (Upper liquid II) is saved for analysis. All fractions are analyzed in the way described in example 1, and the results appear from the following table 3.

Table 3

Fraction	Vol. mls	Protein mg/ml	Total protein mg	protein %	VIII:C g/ml	Total VIII:C g	Specific activity g/mg
Cryo	56	30.4	1702	100	4.29	240	0.14
Upper liquid I	228	-	-	-	0.25	57	-
F I	40	21.1	844	49.6	5.96	238	0.28
Upper liquid II	168	-	-	-	0.97	163	-
F II	30	23.5	705	41.4	0.49	15	0.02
Upper liquid III	198	-	-	-	0.04	8	-
F III	6.2	13.4	83	4.9	27.8	172	2.07

Example 8. In the previous examples, the analyses of the factor VIII complex have only comprised factor VIII:C. In the following table 4 analyses of a factor VIII complex is shown, which has been isolated substantially  
5 in the way described in example 7.

Table 4

Fraction	Prot.	VIII:C	VIII:CAG	F VIII:RAG	VIII:RCF	Spec. activity
Cryo	100	100	100	100	100	0.15
F III	4	63	61	79	48	2.1

Remark: The factor VIII:C can be determined according to Nilsson et al: Acta Med. Scan. 159, 35 (1953) or according to Savidge et al: Thromb. Res. 16, 355 (1979). The factor VIII:CAG can be determined according to Lazarchick and  
10 Hoyer, J. Clin. Invest. 62, 1048 (1978).

It is apparent from the table that the complex, expressed in percent, contains the same concentration of coagulation factor VIII:C as of factor VIII:CAG. The content of factor VIII:RAG is somewhat higher and the content  
15 of factor VIII:RCF (lacking in v. Willebrand's disease) is somewhat lower than of factor VIII:C/CAG.

It is evident that products with a high specific activity and a high solubility can be obtained in a good yield at fractionation with glycine according to the invention  
20 in combination with other fractionating processes. Furthermore, these preparations contain all the activities which in addition to F VIII:C are associated with the factor VIII complex. In comparison with plasma (the specific VIII:C activity of which is about 0.014 U/mg), the specific activity of the product  
25 will be 150-250 times greater, the yield from plasma under optimal conditions being 40-60 %.



CLAIMS

1. A process in purification and/or concentration of factor VIII complex, starting from a preparation of factor VIII complex obtained and concentrated in a known manner, most often in the form of a precipitate, c h a r a c t e r i z e d in that the preparation is dissolved in a glycine solution of at least 1.5 M at a temperature of at least +15°C and a pH of 6.3-7.8, and that a supernatant liquid is recovered as a product or for further working-up.

2. The process of claim 1, c h a r a c t e r i z e d in that it is started from a cryoprecipitate or from Cohn's fraction I-0.

3. The process of claim 1 or 2, c h a r a c t e r i z e d in that the ionic strenght of the solution is 0.1-0.5.

4. The process of any one of claims 1-3, c h a r a c t e r i z e d in that the temperature is between 20 and 40°C, preferably 25-37°C.

5. The process of any one of claims 1-4, c h a r a c t e r i z e d in that pH is adjusted to 6.5-7.5, preferably to about 6.8, before the precipitation with glycine solution.

6. The process of any one of claims 1-5, c h a r a c t e r i z e d in that the glycine solution is a saturated solution, preferably a solution of 1.8-2.2 M, such as about 2.0 M.

FIG.1

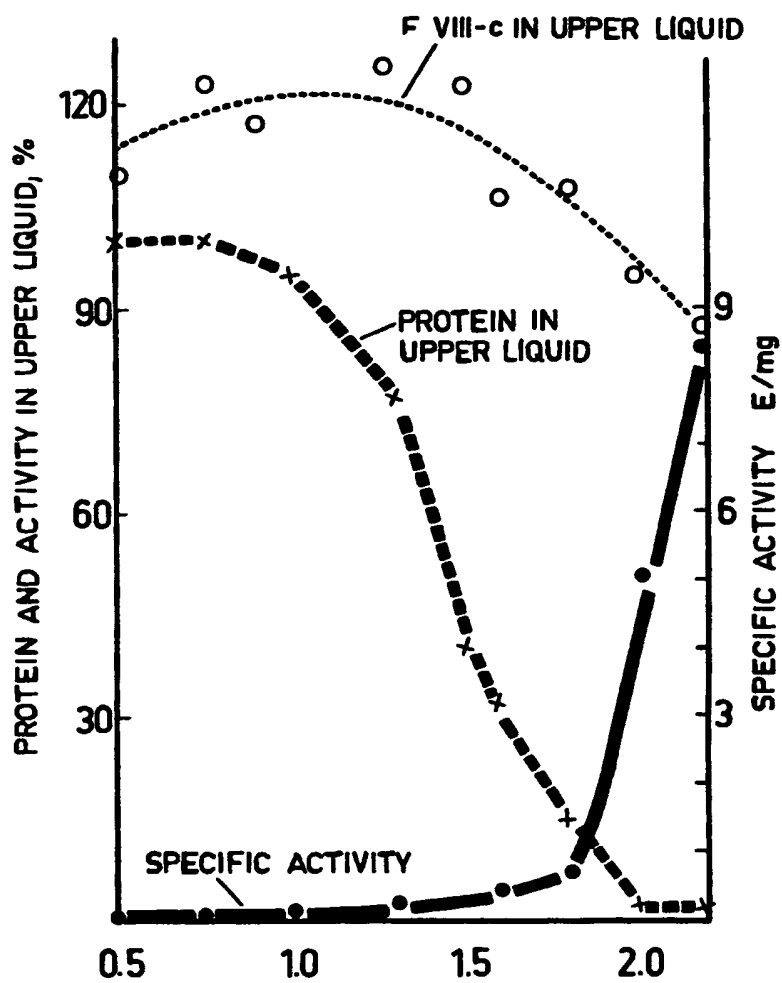


FIG. 2

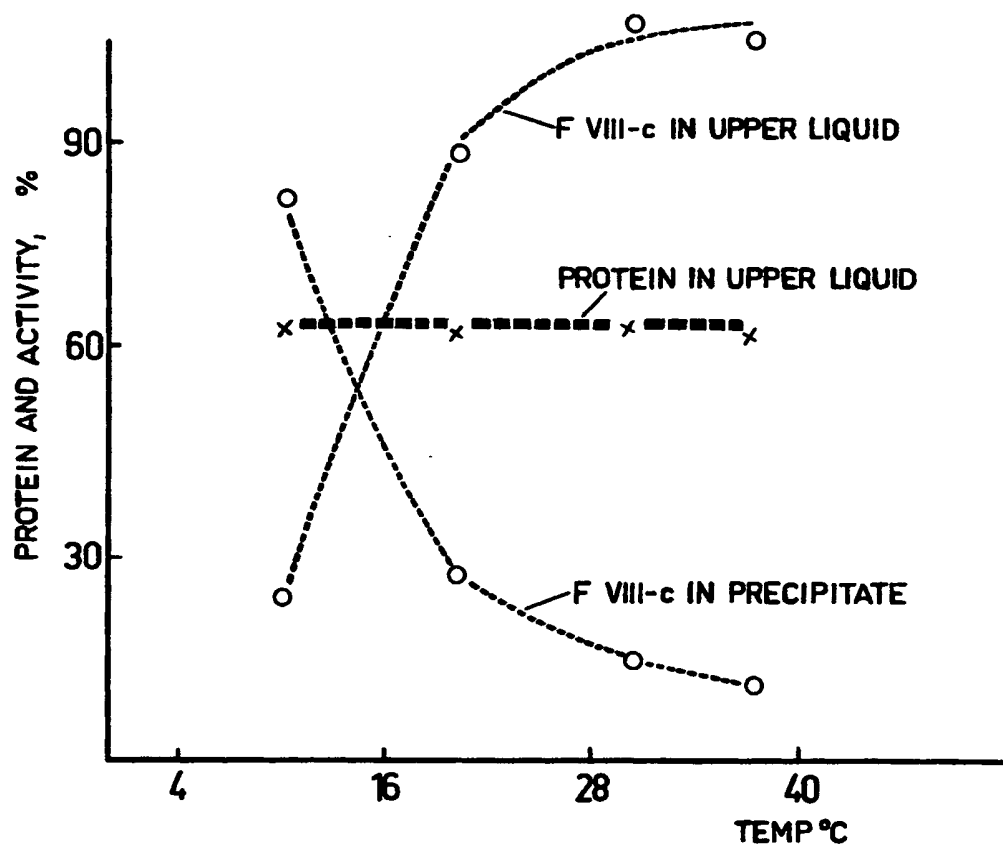


FIG. 3

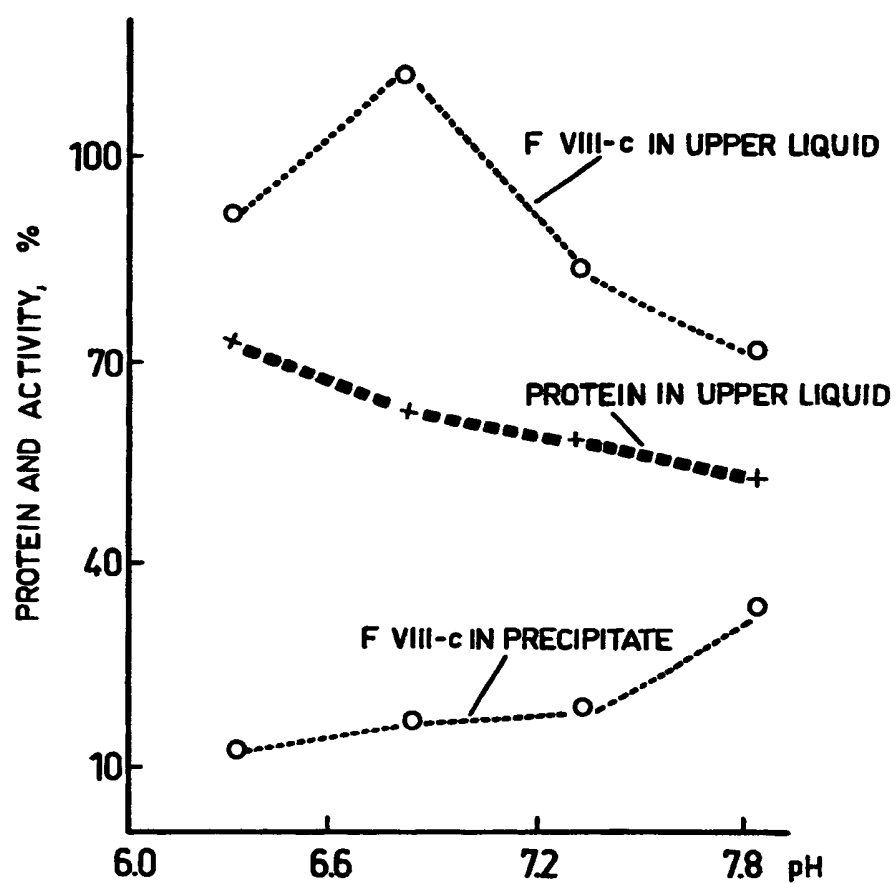
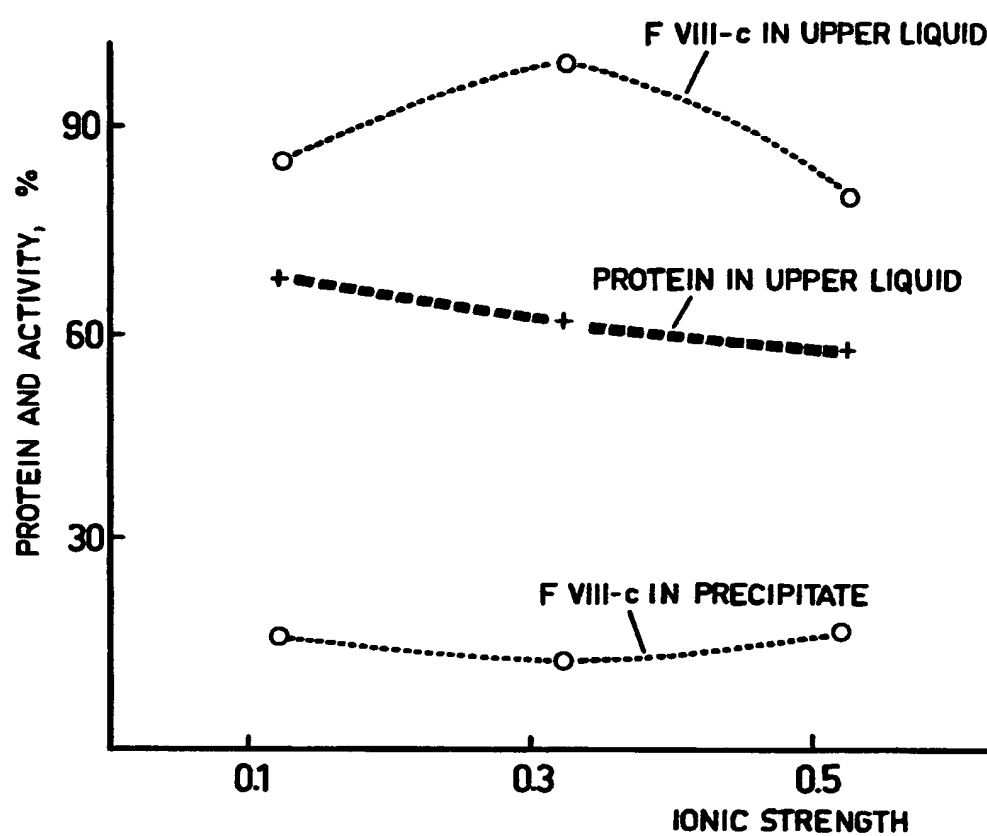


FIG.4





European Patent  
Office

# EUROPEAN SEARCH REPORT

0032655  
Application number

EP 80 85 0193.6

DOCUMENTS CONSIDERED TO BE RELEVANT			CLASSIFICATION OF THE APPLICATION (Int. Cl.3)
Category	Citation of document with indication, where appropriate, of relevant passages	Relevant to claim	
A	<p><u>DE - A1 - 2 854 381</u> (G.A. ROCK) * claims 1, 8 * &amp; US - A - 4 203 891</p> <p>—</p> <p><u>DE - A1 - 2 703 620</u> (RECHERCHES HEMA- TOLOGIQUES) * claim 1; example 1 * &amp; GB - A - 1 557 609</p> <p>—</p> <p><u>US - A - 3 920 625</u> (L.-O. ANDERSSON et al.) * claim *</p> <p>—</p> <p>Chemical Abstracts vol. 85, no. 8, 1976 Columbus, Ohio, USA CUTTER LABORATORIES "Blood coagulation factors" page 337, column 2, abstract no. 51728f &amp; JP - A - 74 - 86515</p> <p>—</p> <p><u>US - A - 3 631 018</u> (E. SHANBROM et al.) * claim 1 *</p> <p>—</p>	<p>1</p> <p>1-5</p> <p>2</p> <p>2</p> <p>1</p>	<p>C 07 G 7/00 C 12 N 9/48</p> <p>TECHNICAL FIELDS SEARCHED (Int. Cl.3)</p> <p>C 07 G 7/00 C 12 N 9/48</p> <p>CATEGORY OF CITED DOCUMENTS</p> <p>X: particularly relevant A: technological background O: non-written disclosure P: intermediate document T: theory or principle underlying the invention E: conflicting application D: document cited in the application L: citation for other reasons</p> <p>&amp;: member of the same patent family, corresponding document</p>
X The present search report has been drawn up for all claims			
Place of search Berlin		Date of completion of the search 03-04-1981	Examiner KNAACK